

## Tris(quinolin-8-olato)gallium(III)

Manju Rajeswaran\* and  
Viktor V. JarikovEastman Kodak Company, Research and  
Development Laboratories, Rochester,  
NY 14650-2106, USACorrespondence e-mail:  
manju.rajeswaran@kodak.com

## Key indicators

Single-crystal X-ray study

 $T = 293\text{ K}$ Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$  $R$  factor = 0.050 $wR$  factor = 0.117

Data-to-parameter ratio = 14.7

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The title compound ( $\text{GaQ}_3$ ),  $[\text{Ga}(\text{C}_9\text{H}_6\text{NO})_3]$ , is the Ga analog of the most widely used organic light-emitting diode (OLED) material,  $\text{AlQ}_3$ . Its molecular structure is a six-coordinated gallium compound, with the angles around the  $\text{Ga}^{3+}$  ion indicating approximate octahedral geometry. There is intermolecular  $\pi$ - $\pi$  stacking of the ligands (8-hydroxyquinolines) in a multidirectional fashion.

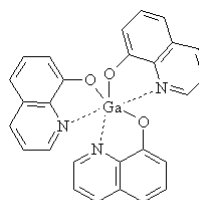
Received 8 December 2003

Accepted 16 January 2004

Online 23 January 2004

## Comment

Quinolinolates of the elements of Group IIIB (denoted  $\text{MQ}_3$ ), Al, Ga, and In, have been of continuous interest to organometallic and physical chemists, in particular, for the last 50 years. Organic light-emitting diodes (OLEDs) utilizing  $\text{MQ}_3$  were first explored in the early 1980s (Tang & Van Slyke, 1985) and continue to be the subject of current research. In 1987, efficient electroluminescence from an OLED device using low molecular-weight organic materials was first reported (Tang & Van Slyke, 1987). The OLED device was constructed of two active layers and used a metal quinolinolate, tris(quinolin-8-olato)aluminum. This discovery generated renewed interest in metal quinolinolates.  $\text{GaQ}_3$  and  $\text{InQ}_3$  are the Ga and In analogs, respectively, of the most widely used OLED material,  $\text{AlQ}_3$ . In our continuing effort to understand how molecular configuration, packing, and polymorphism affect charge transport, electroluminescence evolution, operational stability, and device performance parameters for this series of metal quinolinolates, we report here the single-crystal structure of gallium tris(quinolin-8-olate),  $\text{GaQ}_3$ , (I).



(I)

Tris-chelate quinolin-8-olate metal complexes can occur in two different geometrical forms, facial or meridional. The crystal structure of (I) is a meridional form of  $\text{GaQ}_3$ . The molecular structure of  $\text{GaQ}_3$  (Fig. 1) is a six-coordinated gallium compound. The angles around the  $\text{Ga}^{3+}$  ion indicate approximate octahedral geometry. The average Ga—O and Ga—N distances are 1.931 (3) and 2.091 (3) Å, respectively. These are comparable with those obtained for the solvated  $\text{GaQ}_3$  structure (Wang *et al.*, 1999).

There is intermolecular  $\pi$ - $\pi$  stacking of the ligands (8-hydroxyquinolines) in a multidirectional fashion. Such

intermolecular stacking was also observed in the ethyl ether solvated  $\text{GaQ}_3$  structure (Wang *et al.*, 1999; Brinkmann *et al.*, 2000). The ligands in  $\text{GaQ}_3$  are stacked with an interplanar distance of 3.385–3.460 Å, which is comparable with the values of 3.35–3.41 Å for ether-solvated  $\text{GaQ}_3$  and 3.406–3.428 Å for the  $\text{InQ}_3$  structure (Rajeswaran & Jarikov, 2003).

## Experimental

$\text{GaQ}_3$ , (I), was prepared according to established methods (Lytle *et al.*, 1973). The compound was chemically purified by repeated washings and recrystallizations and subjected to vacuum temperature-gradient sublimation, three consecutive times, until a purity of 99.9% was achieved. The purity of  $\text{GaQ}_3$  was determined by NMR in *d*<sub>7</sub>-dimethylformamide solutions because using  $\text{CD}_2\text{Cl}_2$  and *d*<sub>6</sub>-dimethylsulfoxide resulted in poor peak resolution, while the use of  $\text{CDCl}_3$  produced decomposition (two types of new quinolinol structures were formed).  $\text{GaQ}_3$  was sublimed at 0.6 Torr (1 Torr = 133.322 Pa) and the temperature was gradually increased from 533 to 573 K over a period of 1–3 d. HPLC, ESI LC–MS, and MS show results consistent with the structure. We note that, although the remaining 0.1% impurities could not be positively characterized, they may be different forms, *e.g.*, isomers, of the target compound.

### Crystal data

$[\text{Ga}(\text{C}_9\text{H}_6\text{NO})_3]$	$Z = 2$
$M_r = 502.16$	$D_x = 1.558 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 8.4250 (3) \text{ \AA}$	Cell parameters from 6837 reflections
$b = 10.2900 (3) \text{ \AA}$	$\theta = 1.0\text{--}26.7^\circ$
$c = 13.1390 (5) \text{ \AA}$	$\mu = 1.32 \text{ mm}^{-1}$
$\alpha = 71.4320 (12)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 82.6670 (12)^\circ$	Block, yellow
$\gamma = 89.7690 (14)^\circ$	$0.32 \times 0.25 \times 0.20 \text{ mm}$
$V = 1070.13 (6) \text{ \AA}^3$	

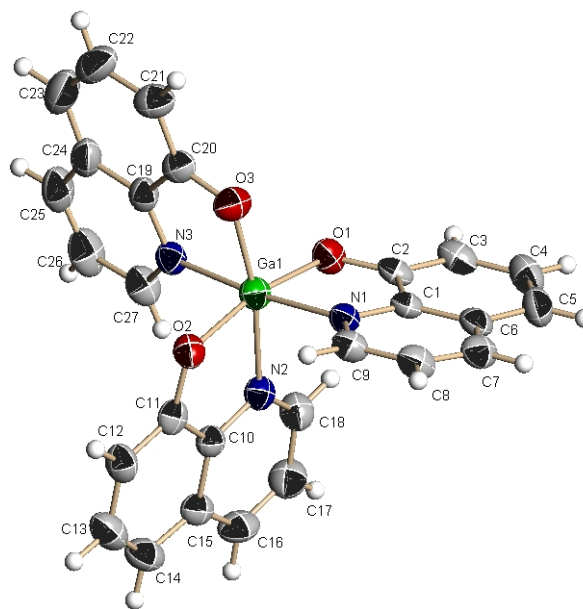
### Data collection

Nonius KappaCCD area-detector diffractometer	4527 independent reflections
$\varphi$ and $\omega$ scans	2412 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	$R_{\text{int}} = 0.144$
$T_{\text{min}} = 0.557$ , $T_{\text{max}} = 0.768$	$\theta_{\text{max}} = 26.8^\circ$
16 177 measured reflections	$h = -10 \rightarrow 10$
	$k = -13 \rightarrow 13$
	$l = -16 \rightarrow 16$

### Refinement

Refinement on $F^2$	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.050$	$w = 1/[\sigma^2(F_o^2) + (0.1357P)^2]$
$wR(F^2) = 0.117$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.89$	$(\Delta/\sigma)_{\text{max}} = 0.001$
4527 reflections	$\Delta\rho_{\text{max}} = 0.41 \text{ e \AA}^{-3}$
307 parameters	$\Delta\rho_{\text{min}} = -0.61 \text{ e \AA}^{-3}$

The positional parameters of the H atoms were calculated geometrically (C–H distances fixed at 0.96 Å) and refined using a riding model ( $U_{\text{iso}} = 1.2U_{\text{eq}}$  of the parent atom). In the final difference Fourier map, the deepest hole is 0.97 Å from the Ga atom. The quality of the  $\text{GaQ}_3$  crystals was not excellent, as indicated by a rather



**Figure 1**

A view of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

high  $R_{\text{int}}$  value of 0.144. There was a minor twin component in the crystals, which was left untreated.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

## References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–58.  
 Brinkmann, M., Gadret, G., Muccini, M., Taliani, C., Masciocchi, N. & Sironi, A. (2000). *J. Am. Chem. Soc.* **122**, 5147–5157.  
 Bruker (2001). *SHELXTL*. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Lytle, F. E., Storey, D. R. & Juricich, M. E. (1973). *Spectrochim. Acta A*, **29**, 1357–1369.  
 Nonius (2000). *COLLECT*. Nonius BV, Delft, The Netherlands.  
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.  
 Rajeswaran, M. & Jarikov, V. V. (2003). *Acta Cryst.* **E59**, m306–m307.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Tang, C. W. & Van Slyke, S. A. (1985). Unpublished results.  
 Tang, C. W. & Van Slyke, S. A. (1987). *Appl. Phys. Lett.* **51**, 913–915.  
 Wang, Y., Zhang, W., Li, Y., Ye, L. & Yang, G. (1999). *Chem. Mater.* **11**, 530–532.